



Armed Forces College of Medicine AFCM



CARBON DIOXIDE TRANSPORT

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INTENDED LEARNING OBJECTIVES (ILOs)



By the end of this lecture the student will be able to:

1. List forms in which carbon dioxide is transported in the blood.
2. Illustrate the carbon dioxide dissociation curve: shape and shifts
3. List the typical values of carbon dioxide content and partial pressures in blood
4. Describe Gas movement in pulmonary and systemic capillaries; chloride shift
5. Compare Bohr and Haldane effects

CARBON DIOXIDE (CO₂) TRANSPORT



- Arterial blood contain 48 ml CO₂ / 100 ml blood at a tension of 40 mmHg
- Venous blood contain 52 ml CO₂ / 100 ml blood at a tension of 45 mmHg.

So every 100 ml blood carries 4 ml from tissues which known as **tidal CO₂**

- **Tidal CO₂**

It is the amount of CO₂ added by the tissues to every 100 cc of arterial blood to be changed into Venous blood (4 ml of

CARBON DIOXIDE (CO₂) TRANSPORT



1) Physical 7% dissolved in plasma

2) Chemical 93%:

a) Carbamino compounds 23%

CO₂ combined with Hb and plasma proteins

b) HCO₃⁻ 70%.

CO₂ in the blood reacts with water to form carbonic acid.

Carbonic anhydrase enzyme which is found in many cells including RBCs, accelerates the reaction:

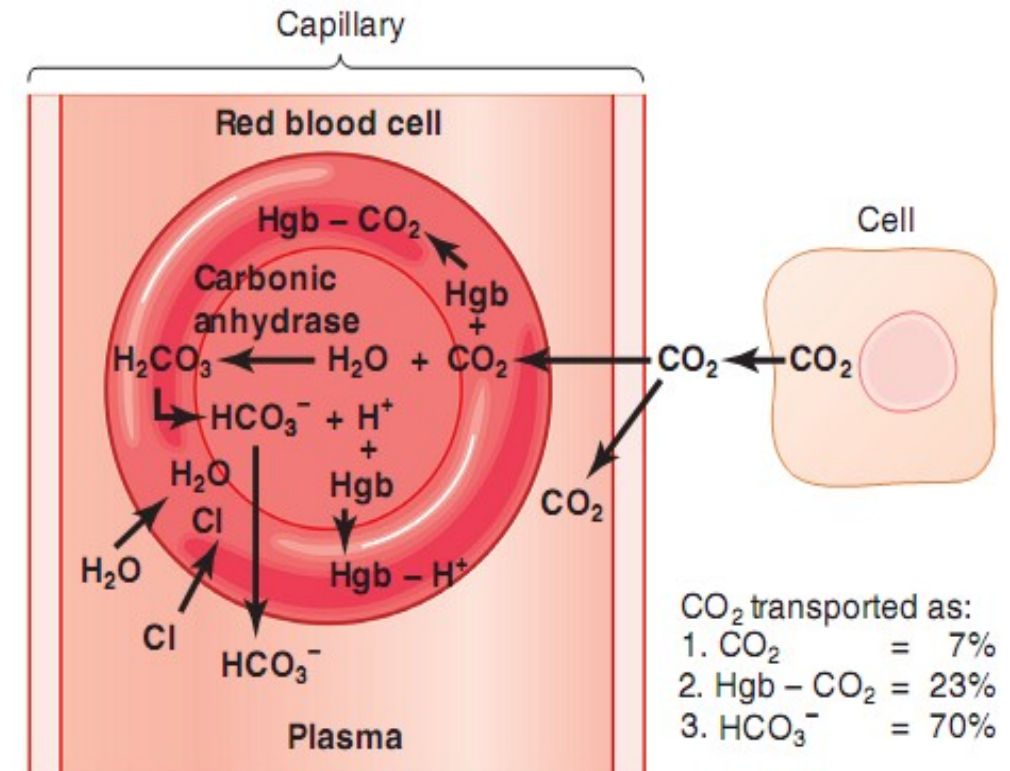


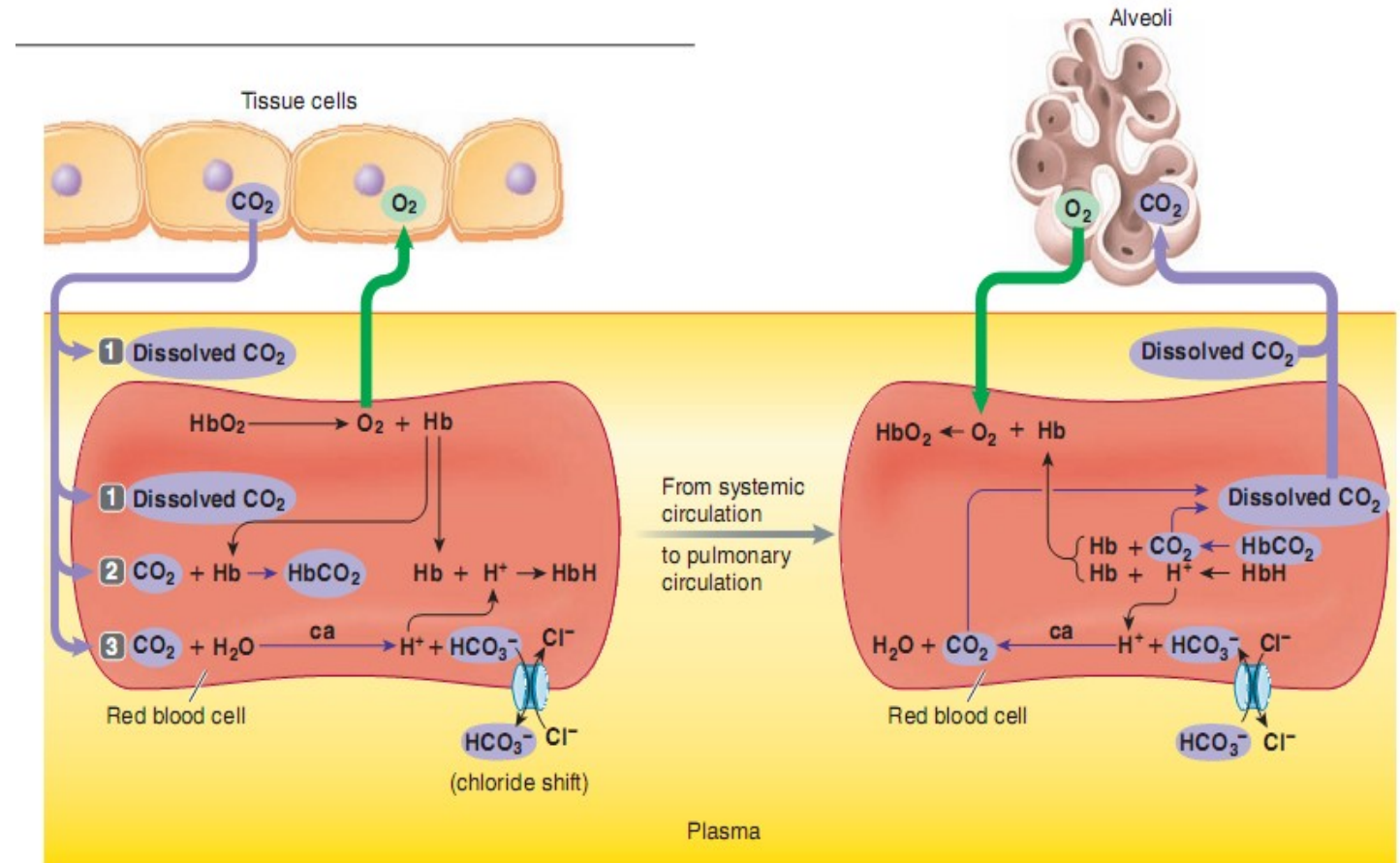
Figure 41-13. Transport of carbon dioxide in the blood.

Guyton and Hall,
2016

CHLORIDE SHIFT



Movement of Cl^- ions into or out of RBCs, to compensate for the movement of HCO_3^- ions to maintain electrical neutrality.



ca = Carbonic anhydrase

Lauralee Sherwood,
2016

Effect of chloride shift



CO_2 entering the blood is converted to HCO_3^- in RBCs by carbonic anhydrase enzyme, most of this HCO_3^- moves out of RBC into plasma in exchange for Cl^- to maintain electrical neutrality.

So net result of Cl^- shift:

3 substances, which increase in both RBCs and plasma:

1- CO_2 2- Carbamino compounds 3- HCO_3^- .

2 substances that increase in RBCs and decrease in plasma:

1- Cl^- 2- H_2O .

So *hematocrit value* is **more** in *venous blood* due to increase size of RBCs.

PHYSIOLOGICAL CO₂ DISSOCIATION CURVE



It represents the relationship between total CO₂ content and PCO₂

It is the line which connects between:

Point A (Venous blood):

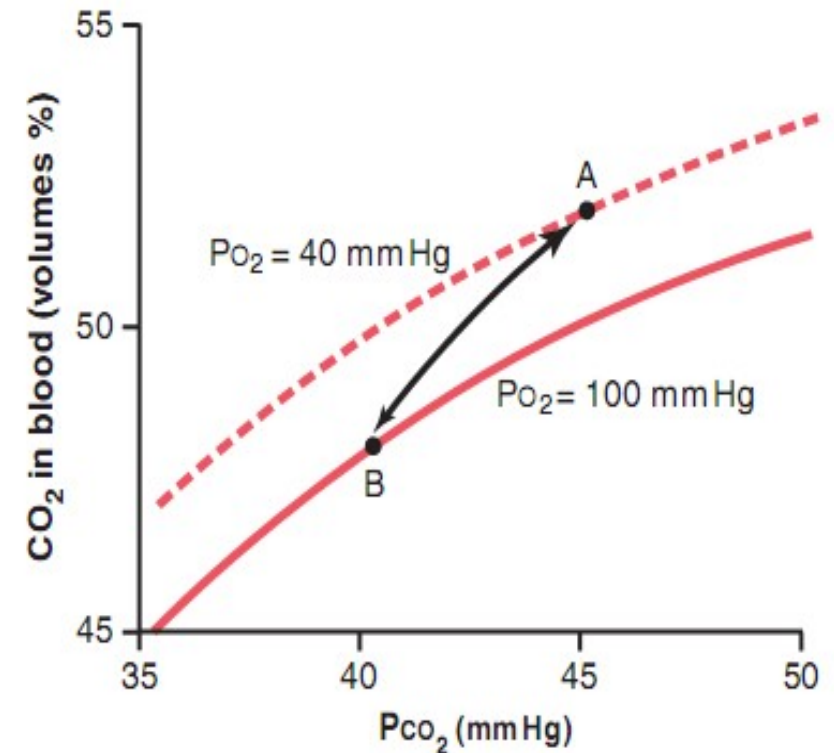
PCO₂ = 45 mmHg

CO₂ content = 52cc

Point B (arterial blood):

PCO₂ = 40 mmHg

CO₂ content = 48cc

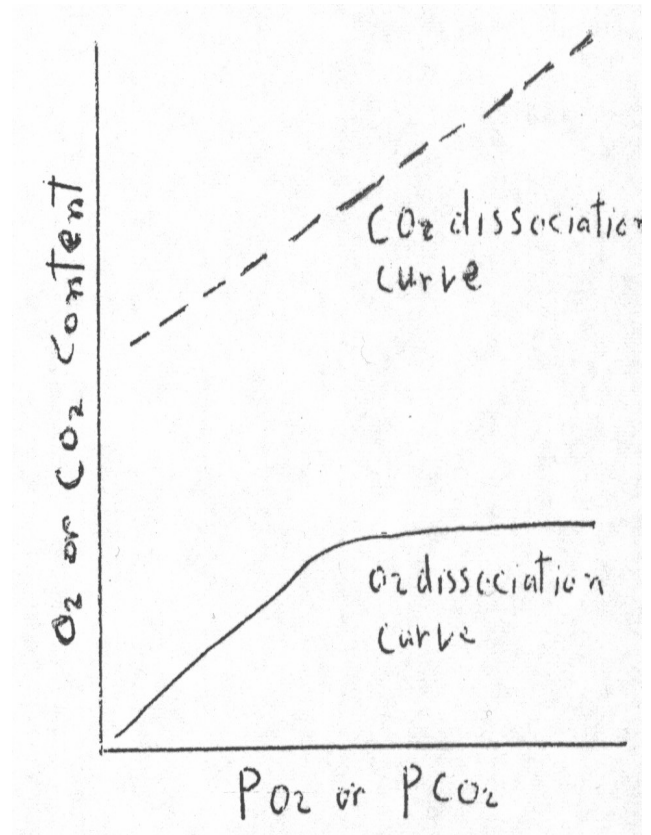


Guyton and Hall,
2016

Comparison between O₂ and CO₂ curves:



- CO₂ content curve is **linear**.
- The CO₂ content of blood is **more than twice** the O₂ content of the blood because of greater solubility in blood.
- CO₂ curve is much **steeper** i.e. much larger changes occur for the same changes in partial pressure.
- Over-ventilation of parts of lung can remove CO₂ to compensate for the under-ventilation of other parts of lungs (such **compensation** can not occur for O₂).



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O_2 , CO_2 and H^+ interaction



Bohr effect:

It is the effect of CO_2 and H^+ on Hb binding with O_2
Binding of CO_2 and H^+ with Hb will decrease Hb affinity to O_2 ,
Shifting of oxyhaemoglobin curve to the right

Haldane effect:

It is the effect of O_2 on Hb binding with CO_2 and H^+
Increasing Hb oxygen (HbO_2) saturation decreases blood CO_2
binding capacity.



1. The main form of carbon dioxide transported in the blood is:

- A. Physically dissolved in plasma.
- B. Carbamino compound.
- C. Carbonic acid.
- D. Bicarbonate.



2. The net result of chloride shift is:

- A. CO_2 increases in plasma & decreases in RBCs .
- B. Electrical neutrality of RBCs is disturbed.
- C. Haematocrit value is more in venous blood.
- D. Carbamino compounds are decreased in RBCs.
- E. Decrease HCO_3^- in RBCs and Plasma

SUGGESTED TEXTBOOKS



1. Guyton and Hall textbook of medical physiology, thirteenth edition 2016, Elsevier, chapter 41 , from page 534 to 536
2. Ganong's Review of Medical Physiology, twenty-fifth edition 2016, McGraw-Hill Education, chapter 35, from page 643 to 641
3. Lauralee Sherwood Human Physiology: From Cells to Systems, Ninth edition 2016. CENGAGE, chapter 13, from page 476 to 477

Thank You